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of each element is represented on the reactant and product sides. Equations must be balanced to accurately reflect the law of conservation of matter.

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ANSWER Answer the following questions in the space provided. 1.

Assign the oxidation number to the specified element in each of the following examples: 4 a. S in H_2SO_3 6 b. S in MgSO_4 2 c. S in K_2S 1 d. Cu in Cu_2S 6 e. Cr

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in Na_2CrO_4 5 f. N in
 HNO_3 4 g. C in $(\text{HCO}_3)_2$
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Answer: (b) Lewis
Base. Explanation:
R⁺NH₂ (Amines)
behaves as a Lewis base

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16. What is the

kinetically controlled

product in the following

reaction? A) A B) B C)

C D) D Ans: A 17. Rank

the following

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Carbocations in
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